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## Chapter 3 Review

Atoms of a given element are identical in size, mass and other properties; atoms of different elements differ in size, mass, and other properties. 3. Atoms cannot be subdivided, created or destroyed. 4. Atoms of different elements combine in simple whole-number ratios to form chemical compounds. 5.

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conservation of mass. mass is neither created  
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reactions or physical changes  $2H_2 + O_2 = 2H_2O$ .  
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chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. The energy required to remove one electron from an atom is called its \_\_\_\_\_. 18.

## Modern Chemistry Chapter 3 Test B Answers

CHAPTER 5 REVIEW The Periodic Law SECTION 1  
SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c)

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according to increasing atomic number. (d)  
based on when they were discovered. 2. d  
Mendeleev noticed that certain similarities  
in the ...

## 5 The Periodic Law

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State the following measured quantities in the units indicated.

- a. 5.2 cm of magnesium ribbon in millimeters
- b. 0.049 kg of sulfur in grams
- c. 1.60 mL of ethanol in microliters
- d. 0.0025 g of vitamin A in micrograms
- e. 0.020 kg of tin in milligrams
- f. 3 kL of saline solution in liters

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