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1. The solubility of a solute is the maximum mass of the solute that will dissolve in a certain amount of water at a certain \_\_\_\_\_. This is the same as saying that solubility is the concentration of a (unsaturated, saturated) solution of the solute. 2. From your graph, what mass of KCl can be dissolved in 100 g of water at these temperatures?

### Mass of KCl ? g KCl = 100 g H<sub>2</sub>O

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