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5070/02/M/J/04 [Turn over A2 Two isotopes of potassium are $^{39}_{19}\text{K}$ and $^{40}_{19}\text{K}$. (a) Complete the table about the number of particles found in one atom of each of these isotopes. [2]

(b) Potassium reacts with water as shown in the equation. $2\text{K}(\text{s}) + 2\text{H}_2\text{O}(\text{l}) \rightarrow 2\text{K}^+(\text{aq}) + 2\text{OH}^-(\text{aq}) + \text{H}_2(\text{g})$ Describe what you would see when potassium reacts with water.

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