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1. How is activation energy related to chemical reactions? All chemical reactions need a certain amount of activation energy to get started. 2. Can you identify if a reaction is exothermic or endothermic based on the graph of activation energy? 3. What factors affect the rate of a chemical reaction? How?

chem study guide answer key | Solution | Chemical ...

the mass of the products of a chemical reaction. is always the same as the mass of the reactants. in that reaction. The atoms have just changed. partners to form new chemical bonds. 8. It is a reaction that releases energy. 9. It is a reaction that absorbs energy. 10. a. Yes. b. No. c. Yes. d. No. Section 2 (page 82) Directed Reading for Content Mastery. Overview (page 83)

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Chemical reactions. When a chemical reaction occurs, the original substances put together, called reactants, lose their chemical properties and become different substances called products with a different set of chemical properties. Reactions where energy is released are called exothermic reactions. When energy is absorbed, it is called an endothermic reaction.

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is a purposeful execution of chemical reactions to obtain a product, or several products. This happens by physical and chemical manipulations usually involving one or more reactions. Unlock all answers Please join to get access

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and Study Workbook Guided Reading and Study Workbook ... Neutrons play an important role in chemical reactions. The Role of Electrons (pages 25–28) 13. The space in which the electrons move is huge compared to the space occupied by the . 14. The electrons farthest from the nucleus or most loosely held are called . 15.

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