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Answers 1. Circle the molecules in the

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following list. H₂ HCl P₄ H₂O Fe S₈
CO₂ Ne 2. Circle the compounds in
the following list. H₂ HCl P₄ H₂O Fe
S₈ CO₂ Ne 3. Label the following as
ionic or covalent: HCl - covalent P₄ -
covalent H₂O - covalent NaCl - ionic
Ca₃(PO₄)₂ - ionic CO₂ - covalent BaS
- ionic 4. Predict the charge each

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Molecules and Ions (Homework)

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Answers 1. Circle the molecules in the following list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 2. Circle the compounds in the following list. H₂ HCl P₄ H₂O Fe S₈ CO₂ Ne 3. Label the following as ionic or covalent: HCl - covalent P₄ -

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Answers
covalent H₂O - covalent NaCl - ionic
Ca₃(PO₄)₂ - ionic Page 4/8.

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BaS - ionic 4 ...

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(Homework) 11. The formula for
sodium sulfide is A) NaS. B) K_2S . C)
NaS₂. D) Na₂S. E) SeS. 12. Almost
all the mass of an atom is
concentrated in the nucleus. CHEM

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Atoms, Molecules and Ions

(Homework) 11. The formula for sodium sulfide is A) NaS. B) K_2S . C) NaS_2 . D) Na_2S . E) SeS. 12. Almost all the mass of an atom is

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Answers concentrated in the nucleus. CHEM 1411. Chapter 2. Atoms, Molecules and Ions (Homework) work on the "click and go ...

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Answers

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John Dalton. 1. elements composed of extremely small particles called atoms; 2. atoms of one element are different from atoms of all other elements 3. compounds are composed of atoms of more than one element; 4. chemical reaction involves only the separation, combinations, or rearrangement of

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Answers; not creation or destruction . law
of definite proportions ...

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1.all matter is composed of atoms. 2. atoms of one element cannot be converted into atoms of another

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Answers in a chemical reaction. 3. all atoms of an element have the same number of protons and electrons which determines the chemical behavior of the element.

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Which of the following illustrates the concept of potential energy and which illustrates kinetic energy? (a) water stored in a water tower. (b) A rubber band stretched around a newspaper. (c) A basketball dropping through a net. (d) A car driving down the

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highway. (e) A flashlight battery. 2.
Which of the following processes are
exothermic? Endothermic? (a ...

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1 CHEM 1411. Chapter 8.Molecular
Geometry and Bonding Theories

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(Homework) W CHEM 1411. Chapter 8. Molecular Geometry and Bonding Theories (Homework) W Multiple Choice Identify the choice that best completes the statement or answers the question. _____ 1. The electrons in the outer shell of an atom are involved in bonding. Another name for the ...

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sodium sulfide is A) NaS. B) K_2S . C)

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Answers: D) Na₂S. E) SeS. 12. Almost all the mass of an atom is concentrated in the nucleus. CHEM 1411. Chapter 2. Atoms, Molecules and Ions (Homework) work on the "click and go". Each problem has to be repeated 5 times, to practice. Copy ...

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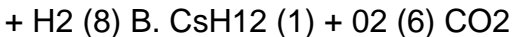
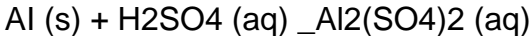
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Balance The Following Equations: A.



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